

## Ch 19 Reaction Rates And Equilibrium Answers

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6 2 3 1 arrhenius equation chemistry libretexts Apr 01 2020 may 29 2021 because a reaction with a small activation energy does not require much energy to reach the transition state it should proceed faster than a reaction with a larger activation energy in addition the arrhenius equation implies that the rate of an uncatalyzed reaction is more affected by temperature than the rate of a catalyzed reaction

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**7 2 rates of reaction and factors affecting rate siyavula Jun 27 2022 it is important to know these factors so that reaction rates can be controlled this is particularly important when it comes to industrial reactions where greater productivity leads to greater profits for companies the following are some of the factors that affect the average rate of a reaction nature of reactants**

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breather this week according to freddie mac the 30 year fixed mortgage rate fell slightly to 6 66  
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speed of a chemical reaction is affected by temperature concentration particle size and the  
presence of a catalyst it can be calculated by measuring changes in reactants  
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